

## Letters to the Editor

### Stereoselective nucleophilic addition of butyllithium and triallylborane to [2.2]paracyclophane-4,7-quinone

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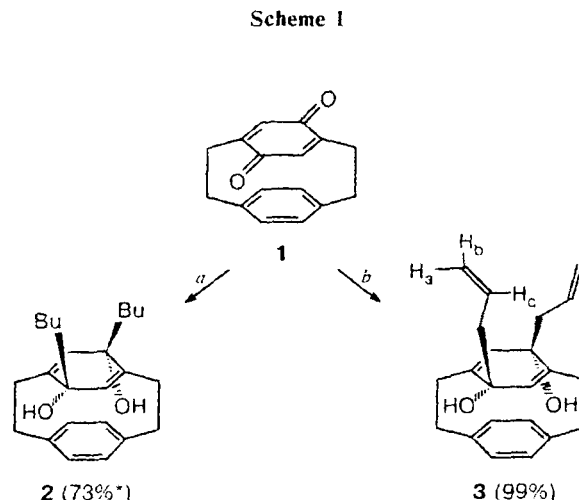
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As part of continuing studies of stereoselective processes in the [2.2]paracyclophane series,<sup>1,2</sup> we carried out experiments on the nucleophilic addition of organometallic reagents to racemic [2.2]paracyclophane-4,7-quinone (**1**).<sup>3</sup>

It is known that the reactions of organomagnesium compounds with *para*-quinones proceed both as 1,2- and 1,4-addition,<sup>4,5</sup> whereas the reactions of organolithium derivatives<sup>4</sup> and triallylborane<sup>6,7</sup> afford only 1,2-addition products. Actually, the reaction of quinone **1** with EtMgI yielded a complex mixture of different hydroxy derivatives, whereas the reaction of compound **1** with BuLi proceeded regio- and stereoselectively to form, according to the <sup>1</sup>H NMR data, *cis*-4,7-dibutyl-4,7-dihydro[2.2]paracyclophane (**2**) in 51% yield, and 30% of the initial quinone **1** was recovered.

The reaction of quinone **1** with AlI<sub>3</sub>B also proceeded stereoselectively to give (after alkaline hydrolysis) *cis*-4,7-diallyl-4,7-dihydroxy-4,7-dihydro[2.2]paracyclophane (**3**) in quantitative yield (Scheme 1).

It should be noted that the reactions of quinone **1** both with BuLi and AlI<sub>3</sub>B afforded diols only as one of the *cis* isomers of two possible *cis* and one *trans* isomer. The <sup>1</sup>H NOESY spectra of compounds **2** and **3**, in which the signals for the OH groups correlate with the signals for the protons of the unsubstituted [2.2]paracyclophane ring, are unambiguously indicative of the *endo*



With respect to the consumed quinone **1**.

**Reagents and conditions:** a. BuLi/PhMe, 20 °C, 10 h;  
b. AlI<sub>3</sub>B/CH<sub>2</sub>Cl<sub>2</sub>, from -78 to 20 °C, 1 h, then MeOH and NaOH.

orientation of the hydroxy groups. Actually, the attack of the nucleophile on the carbonyl groups of quinone **1** from the side which is not shielded by the second [2.2]paracyclophane ring is the most favorable direction, which agrees with the literature data on the addition of

organomagnesium and organolithium reagents to 5-formyl-4-hydroxy[2.2]paracyclophane.<sup>1</sup> As a result, the hydroxy groups in these diols are in the *endo* orientations.

Compounds **2** and **3** were obtained as white crystalline compounds. Both compounds were characterized by mass spectrometry, <sup>1</sup>H NMR spectroscopy, and elemental analysis.

**cis-4,7-Dibutyl-4,7-dihydroxy-4,7-dihydro[2.2]paracyclophane (2).** M.p. 119.5–120 °C. Found (%): C, 81.56; H, 9.74. C<sub>24</sub>H<sub>34</sub>O<sub>2</sub>. Calculated (%): C, 81.31; H, 9.67. MS (EI, 70 eV), *m/z* (*I*<sub>rel</sub> (%)): 354 [M]<sup>+</sup> (0.80); 336 [M – H<sub>2</sub>O]<sup>+</sup> (7.31); 297 [M – Bu]<sup>+</sup> (13.61); 104 (100). <sup>1</sup>H NMR (400.13 MHz, CDCl<sub>3</sub>), δ: 0.83 (t, 6 H, 2 CH<sub>3</sub>, <sup>3</sup>*J* = 7.2 Hz); 1.05–1.50 (m, 12 H, 2 (CH<sub>2</sub>)<sub>3</sub>); 1.90 (s, 2 H, 2 OH); 2.12–2.55 (m, 2 H, bridging CH<sub>2</sub>); 2.65–3.15 (m, 6 H, bridging CH<sub>2</sub>); 4.89 (s, 2 H, H(5), H(8)); 6.96 and 7.05 (both dd, 2 H each, H(12), H(13), H(15), H(16), <sup>3</sup>*J* = 7.8 Hz, <sup>4</sup>*J* = 1.8 Hz).

**cis-4,7-Diallyl-4,7-dihydroxy-4,7-dihydro[2.2]paracyclophane (3).** M.p. 101–101.5 °C. Found (%): C, 81.83; H, 8.00. C<sub>22</sub>H<sub>26</sub>O<sub>2</sub>. Calculated (%): C, 81.95; H, 8.13. MS (EI, 70 eV), *m/z* (*I*<sub>rel</sub> (%)): 304 [M – H<sub>2</sub>O]<sup>+</sup> (1.30); 281 [M – All]<sup>+</sup> (38.34); 263 [M – H<sub>2</sub>O – All]<sup>+</sup> (3.70); 240 [M – 2 All]<sup>+</sup> (14.21); 104 (100). <sup>1</sup>H NMR (400.13 MHz, CDCl<sub>3</sub>), δ: 2.20 (s, 2 H, 2 OH); 1.95–2.05, 2.15–2.30, 2.62–2.73, and 2.83–3.05 (all m, 12 H, two bridging –CH<sub>2</sub>–CH<sub>2</sub>–, two –CH<sub>2</sub>–CH=); 4.90 (s, 2 H, H(5), H(8)); 5.03 (d, 2 H, two H<sub>α</sub>, <sup>3</sup>*J* = 24.1 Hz); 5.10 (d, 2 H, two H<sub>β</sub>, <sup>3</sup>*J* = 9.7 Hz); 5.76 (m, 2 H,

two H<sub>γ</sub>); 6.92 and 7.06 (both d, 2 H each, H(12), H(13), H(15), H(16), <sup>3</sup>*J* = 7.8 Hz).

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## Synthesis of 4,6-dinitro-2-trihalomethyl-2,3-dihydrobenzo[*b*]furans

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A method for the synthesis of previously unknown 4,6-dinitro-2-trihalomethyl-2,3-dihydrobenzo[*b*]furans has been elaborated. The method is based on condensation of 2,4,6-trinitrotoluene with fluoral or chloral in the presence of K<sub>2</sub>CO<sub>3</sub> with subsequent intramolecular cyclization of the resulting 2-picryl-1-(trihalomethyl)ethanols.

**Key words:** 2,4,6-trinitrotoluene, fluoral, chloral, condensation, intramolecular cyclization, 4,6-dinitro-2-trihalomethyl-2,3-dihydrobenzo[*b*]furans.

We found that alcohols obtained by the condensation of 2,4,6-trinitrotoluene (TNT) with fluoral or chloral, viz., 2-picryl-1-(trifluoromethyl)ethanol (**1a**) and 2-picryl-1-(trichloromethyl)ethanol (**1b**), undergo, by the action of bases, intramolecular cyclization (with substitution of the *ortho*-nitro group), which leads to

4,6-dinitro-2-trifluoromethyl-2,3-dihydrobenzo[*b*]furan (**2a**) and 4,6-dinitro-2-trichloromethyl-2,3-dihydrobenzo[*b*]furan (**2b**) (Scheme 1).

The reaction goes smoothly with specified aldehydes or their hydrates; it may serve as a convenient method for preparation of these previously unknown compounds